

Thermo-optical characteristics and concentration quenching effects in Nd³⁺-doped yttrium calcium borate glasses

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(Received 14 January 2011; accepted 24 February 2011; published online 23 March 2011)

In this work we present a comprehensive study of the spectroscopic and thermo-optical properties of a set of samples with composition $x\text{Nd}_2\text{O}_3-(5-x)\text{Y}_2\text{O}_3-40\text{CaO}-55\text{B}_2\text{O}_3$ ($0 \leq x \leq 1.0$ mol%). Their fluorescence quantum efficiency (η) values were determined using the thermal lens technique and the dependence on the ionic concentration was analyzed in terms of energy transfer processes, based on the Förster–Dexter model of multipolar ion–ion interactions. A maximum $\eta = 0.54$ was found to be substantially higher than for yttrium aluminoborate crystals and glasses with comparable Nd³⁺ content. As for the thermo-optical properties of yttrium calcium borate, they are comparable to other well-known laser glasses. The obtained energy transfer microparameters and the weak dependence of η on the Nd³⁺ concentration with a high optimum Nd³⁺ concentration put this system as a strong candidate for photonics applications. © 2011 American Institute of Physics. [doi:10.1063/1.3567091]

I. INTRODUCTION

Yttrium borates in the ternary systems $\text{Y}_2\text{O}_3\text{--Al}_2\text{O}_3\text{--B}_2\text{O}_3$ (YAB) and $\text{Y}_2\text{O}_3\text{--CaO--B}_2\text{O}_3$ (YCOB) are very interesting nonlinear optical crystals with excellent chemical and physical properties.¹ When doped with the trivalent rare earth ions Nd³⁺ and Yb³⁺ they become potential systems for generation of CW and femtosecond-pulsed laser emission in the near-infrared spectral region,^{2–5} and for self frequency doubling in the green.^{6,7} Despite such interesting applications, the requirement of expensive and time consuming techniques for the growth of such crystals has led researchers to explore their glassy counterparts^{8–10} for similar or new applications. Recently, we have reported on the luminescence and thermo-optical properties of Nd³⁺-doped $\text{YAl}_3(\text{BO}_3)_4$ (YAlB) glasses with higher fluorescence quantum efficiency than the Nd:YAB crystal,⁹ and on a detailed structural investigation of the glasses, and their corresponding vitroceraamics, as a function of compositional changes.^{11,12} However, to the best of our knowledge, there have been no reports on the spectroscopic and thermo-optical characterization of undoped and Nd³⁺-doped $\text{YCa}_4\text{O}(\text{BO}_3)_3$ (YCaB) glasses, which have been recently introduced by some of us.^{10,19} A structural study of samples with compositions $x\text{Y}_2\text{O}_3.8x\text{CaO}-(100-9x)\text{B}_2\text{O}_3$ ($x = 4, 5$, and 6 mol%), via Raman and infrared reflectance spectroscopies, has shown that the glasses contain a complex distribution of three-coordinated ortho and pyroborate groups and four-coordinated pentaborate groups, and as the CaO

addition increases there is a clear increase of the former.¹⁰ It was also found that the composition with $x = 5$ is the one that yields the most chemically and thermally stable glasses.

Given the attractiveness of higher flexibility in glass preparation, as compared to crystals, and the possibility of similar and new applications, a full structural and spectroscopic understanding of these new materials is required. With regard to the application as laser media, some very important issues to be addressed are the thermo-optical properties of the host and active ions, and concentration quenching effect due to nonradiative losses via energy migration (EM) and cross relaxations (CRs) between the ions. These features, mainly the latter, can strongly compromise the fluorescence quantum efficiency (η) of the glasses. The experimental determination of the absolute value of η in luminescent solids is known to be a challenging task, due to the requirement of complicated experimental setups, the need of absolute photodetector calibration, the knowledge of absolute ion concentration, and a suitable standard sample. Alternatively, one method that has been successfully used for obtaining η , particularly for Yb³⁺- and Nd³⁺-doped transparent solids, is the normalized-lifetime thermal-lens (TL) method.^{13,14} The method is based on measurements of the photothermal signals and the excited state lifetimes of a set of samples with the same host composition and varying ion concentration. Once η values are determined as a function of ion concentration, they can be effectively employed for the determination of the energy transfer microscopic parameters C_{DX} ($X = D$ for donor–donor transfer as in EM, and $X = A$ for donor–acceptor transfer, as in CRs). Among other systems, we have recently applied this methodology in the study

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TABLE I. Samples and Neodymium concentrations investigated.

Sample	Nd ₂ O ₃ (mol%)	Nd ³⁺ (10 ²⁰ ions/cm ³)
NYCaB10	0.10	0.3898
NYCaB25	0.25	0.9670
NYCaB35	0.35	1.3480
NYCaB50	0.50	1.9110
NYCaB75	0.75	2.8330
NYCaB100	1.00	3.7320

of fluorescence quenching effects in Nd³⁺-doped lead lanthanum zirconate titanate ferroelectric (Nd:PLZT) transparent ceramics for which the results agreed well with theoretical predictions.¹⁵

In this work we present a detailed spectroscopic and thermo-optical characterization of Nd:YCaB glasses with the same atomic ratio of the YCOB crystal ([Y]/[Ca] = 1/4) and 55 mol% boron oxide content. The experimental results are correlated to a theoretical approach,^{13,16–18} aiming to identify and quantify nonradiative contributions responsible for the fluorescence quenching in this system. To this end we employ ground state absorption, steady-state/time-resolved luminescence, and the thermal lens technique in association with the Judd–Ofelt (JO) theory and the Förster–Dexter model for energy transfer.¹⁷

II. EXPERIMENTAL PROCEDURE

The glass samples with composition xNd₂O₃–(5–x)Y₂O₃–40CaO–55B₂O₃ with $0 \leq x \leq 1.0$ mol%, hereafter referred to as YCaB and NYCaB100x, were prepared by the conventional melt quenching method as previously described.^{10,19} Table I presents their corresponding Nd³⁺ concentration in mol% and ions cm^{–3} used in this work. The amorphous nature and structural characterization of the samples were verified by x-ray diffraction and Raman and FT–IR spectroscopies (not shown). In the doping concentration range studied, the structural characteristics of the NYCaB100x samples are very similar to those of the undoped YCaB sample.¹⁰ Ground state absorption was measured in a UV-VIS-NIR Cary 17 spectrometer in the range from 250 to 950 nm, and emission spectra were recorded using a diode laser (810 nm) as excitation source. The luminescent signals were dispersed by a 64 cm single-grated monochromator with 0.1 nm resolution, amplified by a lock-in, and collected by an S-20 photomultiplier. The ⁴F_{3/2} state lifetime values were obtained by analyzing the 1064 nm emission decay curves obtained using a Ge detector with rise time < 3 μs, comparable to the cut-off time of modulation. The thermal lens experiments were carried out in the dual beam mode-mismatched configuration²⁰ using an Ar⁺ laser (514.5 nm) as the excitation source and a HeNe laser (632.8 nm) as the probe beam.

III. RESULTS AND DISCUSSIONS

A. Absorption, luminescence, and Judd–Ofelt analysis

The room temperature absorption coefficient spectrum of the NYCaB75 is presented in Fig. 1. Concerning line posi-

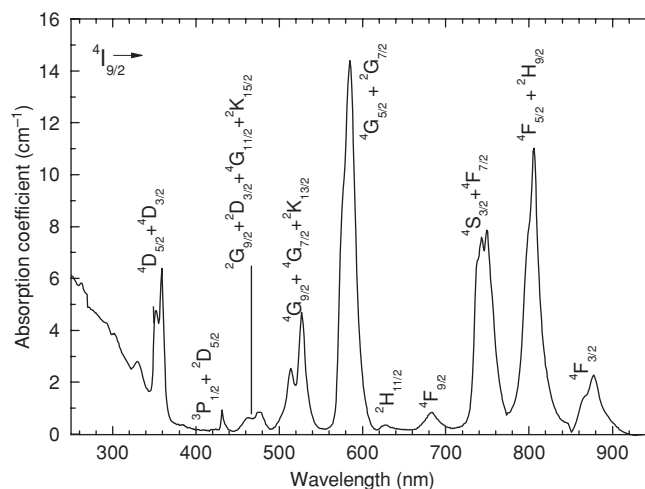


FIG. 1. Representative absorption spectrum of the NYCaB75 glass at room temperature.

tions and widths corresponding to transitions from the ground state ⁴I_{9/2} to the excited states of neodymium, the spectrum is representative for the other samples and resemble that of the Nd:YCOB crystal as well.²¹ The linear dependence of the integrated area of the band at 800 nm (⁴I_{9/2} → ²H_{9/2}, ⁴F_{5/2}) with Nd₂O₃ concentration (not shown) indicates that the ionic incorporation was successful in the range studied, and it is in agreement with the nominal compositions. The UV edge of these glasses is around 240 nm, wider than in bismuth-borate glasses,²² and the energy gap was calculated to be around 5.9 eV,¹⁹ comparable to that of other glass compositions containing up to 80% boron oxide content.^{19,23}

In Fig. 2, the representative luminescence spectrum of the NYCaB75 glass displays the three typical emissions from the emitting level ⁴F_{3/2} to the terminal levels ⁴I_{9/2}, ⁴I_{11/2}, and ⁴I_{13/2} of Nd³⁺, at 900, 1060, and 1320 nm, respectively. As it is common for Nd³⁺-doped systems, the transition to level ⁴I_{15/2} is not detected in this intensity scale due to its very small branching ratio (≈1%). The inset of the same figure presents the integrated intensity for the 1060 nm emission as

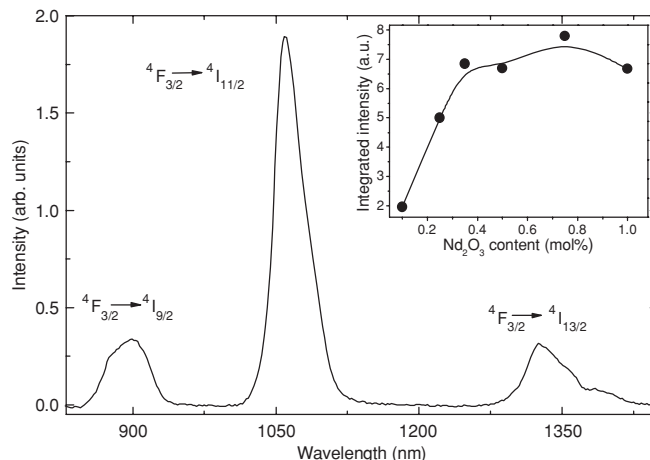


FIG. 2. Luminescence spectrum of the NYCaB75 glass at room temperature and $\lambda_{\text{exc}} = 810$ nm. The inset presents the integrated intensity of the band at 1060 nm as a function of Nd₂O₃ content.

TABLE II. Experimental and calculated Nd³⁺ oscillator strength values for the NYCaB100x glasses.

Sample	Transition (⁴ I _{9/2} → ...)	Wavelength (nm)	$f_{\text{exp}} (\times 10^{-6})$	$f_{\text{cal}} (\times 10^{-6})$	Error (%)
NYCaB10	(1) ⁴ F _{3/2}	875.8	4.892	5.423	10.80
	(2) ⁴ F _{5/2} , ² H _{9/2}	804.0	15.773	15.826	0.34
	(3) ⁴ F _{7/2} , ⁴ S _{3/2}	743.5	15.755	15.826	0.45
	(4) ⁴ G _{5/2} , ² G _{7/2}	579.5	31.662	31.74	0.25
	(5) ⁴ G _{9/2} , ⁴ G _{7/2} , ² K _{13/2}	519.0	13.366	12.073	9.67
NYCaB35	(1) ⁴ F _{3/2}	875.8	4.272	4.788	12.08
	(2) ⁴ F _{5/2} , ² H _{9/2}	804.0	14.401	14.157	1.69
	(3) ⁴ F _{7/2} , ⁴ S _{3/2}	743.5	13.976	14.221	1.75
	(4) ⁴ G _{5/2} , ² G _{7/2}	579.5	31.29	31.337	0.15
	(5) ⁴ G _{9/2} , ⁴ G _{7/2} , ² K _{13/2}	519.0	11.66	10.955	6.05
NYCaB75	(1) ⁴ F _{5/2} , ² H _{9/2}	804.0	14.921	15.021	0.67
	(2) ⁴ F _{7/2} , ⁴ S _{3/2}	743.5	15.308	15.252	0.37
	(3) ⁴ F _{9/2}	680.6	1.277	1.206	5.56
	(4) ⁴ G _{5/2} , ² G _{7/2}	579.5	32.967	32.977	0.03
	(5) ⁴ G _{9/2} , ⁴ G _{7/2} , ² K _{13/2}	519.0	11.683	11.499	1.57

a function of Nd³⁺ concentration showing a maximum for the NYCaB75 sample. For NYCaB samples doped with up to 0.5 mol% the effective width of the 1060 nm band $\Delta\lambda_{\text{eff}} = [\int I_{\text{lum}}(\lambda)d\lambda / I_{\text{lum}}(\lambda)_{\text{peak}}]$ (where $I_{\text{lum}}(\lambda)$ is the emission intensity for the wavelength λ and $I_{\text{lum}}(\lambda)_{\text{peak}}$ is the peak intensity) is 36 nm whereas for the NYCaB75 and NYCaB100 it is 38 nm. This slight, but consistent, increase might be an indication that for these concentrations and higher ones, a fraction of Nd³⁺ ions occupy Ca²⁺ environments in addition to those of Y³⁺. In Nd_xLa_{2-x}B₁₀O₁₉ crystals, inequivalent Nd³⁺ environments were evidenced for $x > 0.036$ (0.25 mol%),²⁴ as well as in Nd³⁺ or Yb³⁺ doped YCOB and GdCa₄O(BO₃)₃ (GdCOB) crystals.^{25,26}

On the basis of the ground state absorption and the luminescence spectra of three samples (0.10, 0.35, and 0.75 mol%), the radiative properties of the NYCaB100x glasses were evaluated via the Judd-Ofelt formalism.²⁷ These parameters were obtained by solving a system of equations from the five main absorption bands, as indicated in Table II, where we compare the experimental and calculated oscillator strengths of these bands. The obtained values of the phenomenological intensity parameters Ω_2 , Ω_4 , and Ω_6 are presented in Table III, with average values of $\Omega_2 = 5.25 \times 10^{-20}$, $\Omega_4 = 9.60 \times 10^{-20}$, and $\Omega_6 = 9.87 \times 10^{-20}$ cm². These values are much larger than those of Nd:YCOB (Ref. 21) and slightly larger than those of Nd³⁺-doped yttrium – aluminoborate (NYAB) glasses and crystals.^{9,28} The average radiative lifetime (τ_{rad}) value of the emitting level ⁴F_{3/2} is 150 μ s (Table IV), relatively smaller than that of Nd:YAIB glass,²¹

TABLE III. Phenomenological JO intensity parameters for NYCaB100x glasses for three Nd³⁺ concentrations.

Sample	$\Omega_2 (\times 10^{-20} \text{ cm}^2)$	$\Omega_4 (\times 10^{-20} \text{ cm}^2)$	$\Omega_6 (\times 10^{-20} \text{ cm}^2)$
NYCaB10	4.66	10.34	10.32
NYCaB35	5.35	9.09	9.29
NYCaB75	5.74	9.37	9.99
rms error = 4%			

half of that for NYAB crystal, and a quarter of that for Nd:YCOB crystal.²¹ It is generally accepted that the Ω_2 parameter is an indicator of the metal–ligand bond covalence whereas the dimension of Ω_6 is related to the rigidity of the host. Following these premises, the fact that Ω_2 is higher for the NYCaB100x glasses than, for instance, for Nd³⁺-doped YAIB (Ref. 9) and Bi₂O₃–B₂O₃ (Ref. 30) glasses is consistent with a higher asymmetry of the ligand field around the Nd³⁺ ions which can, in principle, entail higher M–O orbital superposition. The latter, implies in extended relaxation of selection rules and consequently higher transition probabilities, which would explain the relatively shorter radiative lifetime values of NYCaB100x as compared to Nd:YAIB, for instance. This interpretation however, should not be viewed as a definite one since major structural differences in these systems might play a role. In the case of YAIB, it is known that Nd³⁺ ions can occupy solely Y³⁺ environments,¹¹ but for the YCaB glasses, in which Ca²⁺ also substitutes for Y³⁺, the analysis of the far infrared reflectivity spectra has shown the possibility of different Ca²⁺ occupational environments¹⁰ similar to the case of binary CaO–B₂O₃ glasses and crystals.²⁹ Structural investigations of NYCaB100x by high resolution solid

TABLE IV. Radiative transition rates and branching ratios for NYCaB100x glasses.

Sample	Transition ⁴ F _{3/2} →	$\nu (\text{cm}^{-1})$	Radiative rate (s ⁻¹)	β
NYCaB10	⁴ I _{9/2}	11 111.11	2920.48	0.414
	⁴ I _{11/2}	9442.87	3475.72	0.493
	⁴ I _{13/2}	7490.64	659.21	0.093
				$A_{\text{rad}} = 7055.41$
NYCaB35	⁴ I _{9/2}	11 111.11	2578.79	0.411
	⁴ I _{11/2}	9442.87	3108.57	0.495
	⁴ I _{13/2}	7490.64	593.12	0.094
				$A_{\text{rad}} = 6280.48$
NYCaB75	⁴ I _{9/2}	11 111.11	2679.95	0.405
	⁴ I _{11/2}	9442.87	3307.04	0.499
	⁴ I _{13/2}	7490.64	637.78	0.096
				$A_{\text{rad}} = 6624.77$

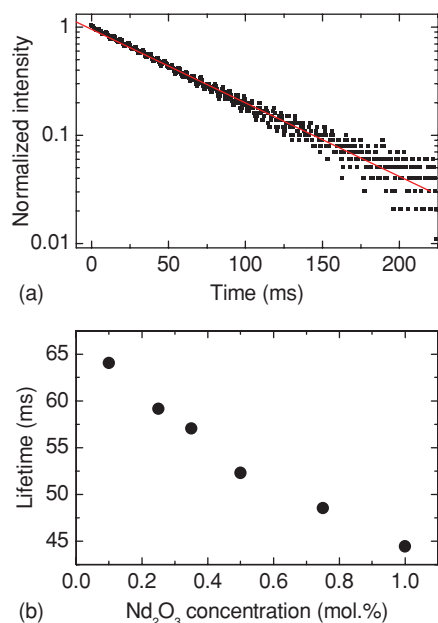


FIG. 3. (a) Luminescence intensity decay profile of the NYCaB25 glass ($\lambda_{\text{exc}} = 800$ nm); (b) Nd_2O_3 concentration dependence for $^4\text{F}_{3/2}$ lifetime values in NYCaB100x glasses.

state NMR are currently being carried out and hopefully will provide further clarifications into such questions.

The experimental lifetime (τ_{exp}) values of the level $^4\text{F}_{3/2}$ were determined from the first decay (e^{-1}) of the luminescence intensity, $I_{\text{lum}}(t)$, and for comparison, using the equation $\tau_{\text{eff}} = [\int I_{\text{lum}}(t)dt]/I_{\text{lum}}(0)$, which presents an effective lifetime, where $I_{\text{lum}}(0)$ is the luminescence intensity for $t = 0$. No appreciable difference, within $\sim 4\%$, was observed in the values obtained through both methods, indicating an exponential behavior to the decay of all the samples. Figure 3(a) shows a typical luminescence intensity decay spectrum for the NYCaB025 sample and Fig. 3(b) the dependence of the lifetime (τ) with the Nd_2O_3 concentration. A typical behavior of τ with the Nd_2O_3 concentration is observed and attributed to interactions between Nd^{3+} ions as CRs and EM.

B. Thermo-optical properties and fluorescence quantum efficiencies

The generation of heat and its effects on the thermo-optical properties of the NYCaB100x glasses were analyzed by the TL technique, in a similar way to our previous report on Nd:YAIB glasses.⁹ The heat deposition is due to the absorption of light followed by nonradiative decays and it manifests itself through the establishment of temperature and refractive index gradients responsible for the lenslike behavior of the samples. This effect is analyzed in terms of the temperature coefficient of the optical path length change (ds/dT). In the dual beam mode-mismatched TL configuration,²⁰ the propagation of a probe laser beam through the lenslike region results in a variation in its on-axis intensity which can be calculated using diffraction integral theory. The temporal evolution of the TL signal depends on the characteristic TL signal time t_c , which is related to the thermal diffusivity

(D) by $D = w_e^2/4t_c$. The latter relates to the thermal conductivity (K) by $K = \rho cD$, where ρ is the sample density and c is the specific heat. The TL transient signal amplitude θ is approximately the phase shift of the probe beam at $r = 0$ and $r = \sqrt{2}w_e$ induced by TL. When normalized by the absorbed power (P_{abs}), θ can be redefined as

$$\Theta = \frac{\theta}{P_{\text{abs}}} = -B\varphi, \quad (1)$$

where $B = (K\lambda_p)^{-1}ds/dT$ is a constant that depends only on the host matrix characteristics and on the probe beam wavelength λ_p , and φ is the fraction of absorbed energy converted into heat. The latter, also called fractional thermal loading, is related to the fluorescence quantum efficiency as

$$\varphi = 1 - \eta \frac{\lambda_{\text{exc}}}{\langle \lambda_{\text{em}} \rangle}, \quad (2)$$

where λ_{exc} is the excitation wavelength and $\langle \lambda_{\text{em}} \rangle$ is the average emission wavelength obtained with the relation $\langle \lambda_{\text{em}} \rangle = \sum \lambda_i \beta_i$, where λ_i is the wavelengths of emissions from the emitting level and β_i is the branching ratios related to such emissions probabilities. Thus, by knowing θ and the host parameters contained in the constant B , one can, in principle, easily obtain η . One problem often encountered, however, is the lack of a reference and undoped sample for the determination of the host characteristics and/or their very low absorption coefficient which makes the TL signal hardly detectable and impedes the precise calculation of the absorbed power. In order to circumvent such problems in the study of NYCaB100x glasses, we employ the alternative method called normalized-lifetime thermal-lens.^{13,14}

Figure 4(a) presents a characteristic TL transient signal for the NYCaB75 sample. The incident power for this transient was 30 mW and the fitting of the experimental

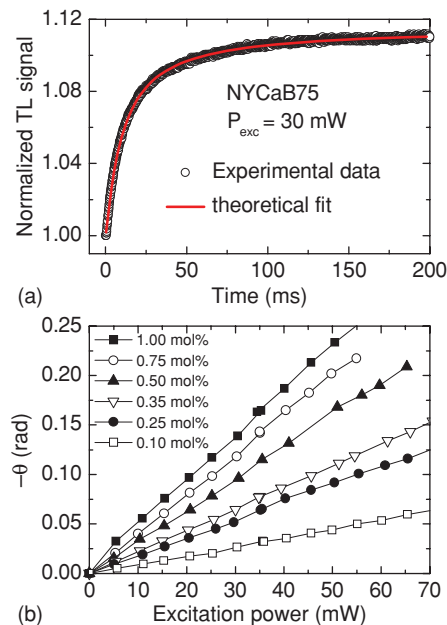


FIG. 4. (a) Representative thermal lens transient signal for NYCaB75 sample with $\lambda_{\text{exc}} = 514.5$ nm and $\lambda_p = 632.8$ nm; (b) Thermal lens phase shift (θ) against the excitation power for NYCaB100x glasses doped with different Nd_2O_3 concentrations.

curve with the TL equation²⁰ (solid line) yields $t_c = (1.18 \pm 0.11)$ ms and $\theta = -(0.118 \pm 0.004)$ rad. The parameters t_c and θ were obtained for all the samples as a function of pump power and doping concentration as shown in Fig. 4(b). In agreement with Eq. (1) a linear dependence of θ with the excitation power P_e is observed. Having obtained t_c , the average value of $D = (3.3 \pm 0.3) \times 10^{-3}$ cm²/s was calculated, and using this parameter together with $\rho = (2.94 \pm 0.03)$ g/cm³ and $c = 0.727$ J/g K (this latter obtained from the Dulong–Petit relation), it was possible to calculate $K = (7.1 \pm 0.7) \times 10^{-3}$ W/cm K. These values are somewhat lower than those determined for Nd:YAIB glasses [$D = (4.1 \pm 0.3) \times 10^{-3}$ cm²/s and $K = (10.6 \pm 0.8) \times 10^{-3}$ cm²/s],⁹ but similar to those of the majority of the laser glasses.²⁰ Once the thermo-optical properties of the host YCaB were determined we proceeded to obtain η for all the doped samples.

Rigorously, the quantum efficiency is defined as the ratio of experimental to radiative lifetime values ($\eta = \tau_{\text{exp}}/\tau_{\text{rad}}$). Since τ_{rad} is independent of the Nd³⁺ concentration, the lifetimes of two samples with concentrations N_0 and N_x , for instance, can be compared as

$$\eta_{N_x} = \eta_{N_0} \frac{\tau_{N_x}}{\tau_{N_0}} = \eta_{N_0} \Gamma(N), \quad (3)$$

where

$$\Gamma(N) = \frac{\tau_{N_x}}{\tau_{N_0}}. \quad (4)$$

Combining Eqs. (1)–(3) the expression for the TL phase shift divided by the absorbed pump power can be rewritten as

$$\Theta = B \left[1 - \Gamma \eta_{N_0} \frac{\lambda_{\text{exc}}}{(\lambda_{\text{em}})} \right], \quad (5)$$

and the lifetime value τ_{N_0} of a chosen sample can be used to normalize the lifetimes values τ_{N_x} of the other samples in the set, so that through the behavior of Θ versus Γ (see Fig. 5) the values of η_{N_0} can be determined through linear fitting, and further used in Eq. (3) to obtain η for the other samples. As the reference sample we chose the lowest doped one (0.1 mol%) for which it was found that $\eta = 0.54$. In addition, the value of $B = (11.9 \pm 0.5) \text{ W}^{-1}$ was also

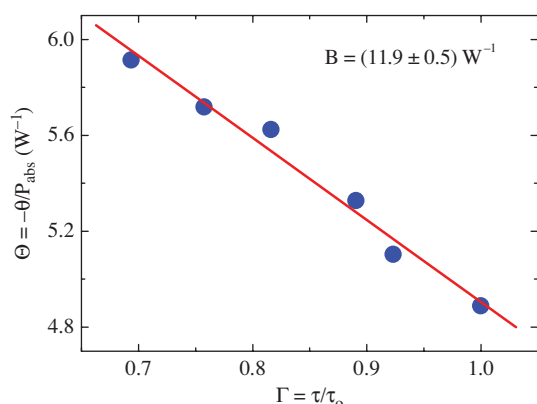


FIG. 5. (dots) Thermal lens phase shift divided by the absorbed pump power against the normalized lifetime $\Gamma = \tau/\tau_0$ for the NYCaB samples. The solid line is the linear fit.

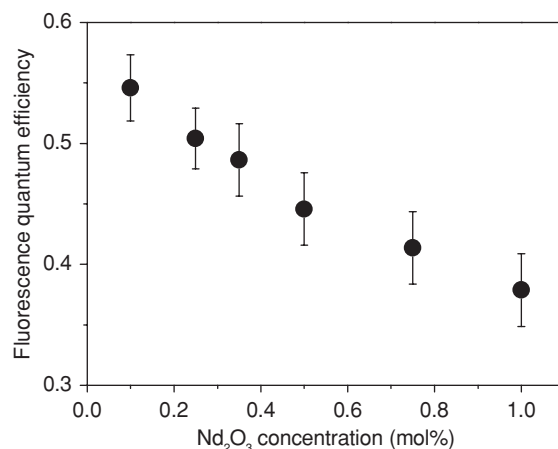


FIG. 6. Fluorescence quantum efficiency of NYCaB100x glasses as a function of Nd₂O₃ concentration.

obtained from the fitting, allowing the calculation of $ds/dT = (5.3 \pm 0.3) \times 10^{-6} \text{ K}^{-1}$. Accordingly, this result is also lower than that of the Nd:YAIB glasses,⁹ half of the aluminosilicate, larger than that of the phosphates, and fluorides as Zr-Ba-La-Al-Na (ZBLAN) glasses.^{13,20} In resume, the D , K , and ds/dT values are typical of laser glasses.

The dependence of η values with Nd₂O₃ concentration is depicted in Fig. 6. It is found that these values are lower than unity but similar to those found in bismuth borate glasses,³⁰ where quasicontinuous laser operation was recently demonstrated.³¹ Moreover, these values are higher than those for Nd³⁺ doped GdCOB crystals ($\eta = 14.8\%$ for $2.2\text{--}8.8 \times 10^{19}$ Nd³⁺ ions/cm³),²⁶ and considerably higher than those for Nd:YAIB glasses⁹ and Nd³⁺ doped YAB crystals²⁸ with comparable Nd³⁺ concentrations. The relatively low value for the less doped sample is attributed to multiphonon decay due to high phonon energy of this matrix, which is from 1100 to 1400 cm⁻¹. The reason for the decrease in η values with increasing concentration is, as earlier mentioned, the growing probability of the energy transfer processes by CR and EM, among Nd³⁺ ions, followed by nonradiative decays (multiphonon in the case of CR). Therefore, it is useful to investigate such contributions.

C. Energy transfer analysis

In general, the emission decay profiles of luminescent host-ion systems encompass radiative and nonradiative contributions. Particularly, the latter can lead to pronounced deviations from a single exponential pattern in lower or greater extent. Thus, profiles such as those in Fig. 3(a) are best described by the following equation:

$$I_{\text{lum}}(t) = I_0 \exp[-(A_{\text{rad}} + W_{\text{mp}})t - \bar{W}t - \gamma\sqrt{t}], \quad (6)$$

where A_{rad} and W_{mp} are the radiative and multiphonon decay rates, respectively, which are independent of ion concentration. \bar{W} accounts for the migration assisted cross relaxation energy transfer^{13,15,16} and $\exp(-\gamma\sqrt{t})$ is the classic Förster decay function,^{13,16,17,32} also known as static disorder decay, accounting for direct cross relaxation. Although both \bar{W} and γ

depend on concentration, only the latter leads to a deviation of the purely exponential behavior of the decay. This fact is usually attributed to the different distances between donors and acceptors leading to a decay rate that differs for each donor–acceptor pair. The resulting overall decay is then nonexponential, with an initial faster decay, due to sites with smaller separation between pairs. According to the Förster model,³² γ is given by

$$\gamma = \frac{4}{3}\pi^{3/2}N_t\sqrt{C_{DA}}. \quad (7)$$

In 1983, Burshtein proposed the “hopping” model¹⁸ to take into account the EM-assisted cross relaxation, through the following equation for \bar{W} :

$$\bar{W} = \pi \left(\frac{2\pi}{3} \right)^{5/2} \sqrt{C_{DA}C_{DD}N_t^2}. \quad (8)$$

In Eqs. (7) and (8), C_{DD} and C_{DA} stand for the microscopic energy transfer parameters that can be calculated using Dexter model of multipolar interactions.¹⁷ The equations are intentionally written for the case of singly doped materials where the concentration of donor and acceptor ions, N_D and N_A , respectively, are equal to N_t , which corresponds to the total Nd^{3+} doping concentration. For this ion, the energy transfer processes can be well described on the basis of dipole–dipole interactions. The transitions involved in the CR related to C_{DA} are ${}^4F_{3/2}$, ${}^4I_{9/2} \rightarrow {}^4I_{15/2}$, ${}^4I_{15/2}$ and/or ${}^4F_{3/2}$, ${}^4I_{9/2} \rightarrow {}^4I_{13/2}$, ${}^4I_{15/2}$; and in the EM related to C_{DD} is ${}^4F_{3/2}$, ${}^4I_{9/2} \rightarrow {}^4I_{9/2}$, ${}^4F_{3/2}$.^{15,16} The microscopic parameters can be obtained by using the spectral overlap integrals of donor ion emission (σ_D^{em}) and acceptor ion absorption (σ_A^{abs}) cross sections according to^{13,15,17}

$$C_{DX} = \frac{3c}{8\pi^4 n_0^2} \int \sigma_D^{\text{em}}(\lambda) \sigma_X^{\text{abs}}(\lambda) d\lambda, \quad (9)$$

where X is equal to D or to A for the EM or CR mechanisms, respectively, c is the speed of light, and n_0 is the refractive index of the medium. The calculation of C_{DD} is straightforward because energy migration is a resonant process and, therefore, the overlap integral has appreciable values. However, the CRs are nonresonant and the obtaining of the overlap integrals for the calculation of C_{DA} requires some data manipulation as to account for the phonon-assistance.¹⁷ This procedure introduces further errors in the calculations, resulting in C_{DA} values that are usually reliable but not as much as that of C_{DD} .

The experimental data in Fig. 6 were also evaluated in terms of the microscopic energy transfer parameters by using Eq. (10) in agreement with Eq. (9).^{13,15,16}

$$\eta(N_t, C_{DA}, C_{DD}) = \frac{A_{\text{rad}}}{I(0)} \int_0^\infty I(t) dt = \frac{A_r}{W_\infty} \times \{1 - \sqrt{\pi} x \exp(x^2) [1 - \text{erf}(x)]\}, \quad (10)$$

where $W_\infty = A_{\text{rad}} + W_{\text{mp}} + \bar{W}$ and $x = \gamma/2\sqrt{W_\infty}$. The $C_{DD} = (4.5 \pm 0.3) \times 10^{-39} \text{ cm}^6/\text{s}$ value was calculated according to Eq. (9). The error bar was estimated by calculating C_{DD} for two samples and using three methods to obtain the emission cross section.^{33,34} After that, the C_{DD} value was used in

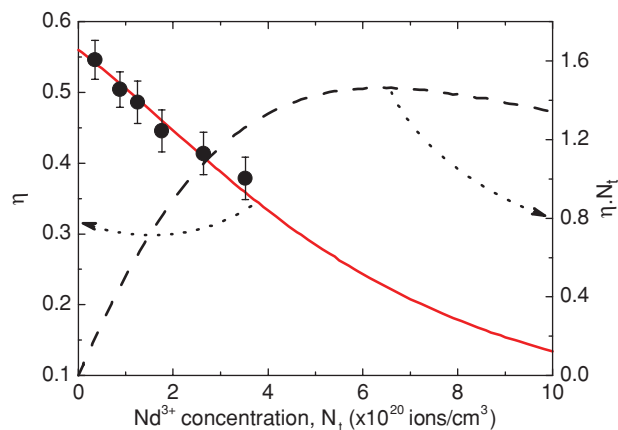


FIG. 7. Dependence on Nd^{3+} concentration of fluorescence quantum efficiency (η) and figure of merit (ηN_t) for ${}^4F_{3/2}$ level in NYCaB100x glass. Open symbols are η results obtained using the thermal lens technique; solid line is the fit of the open symbols using the Eq. (10); and dashed line is the figure of merit calculated with the solid line.

Eq. (10) by letting C_{DA} and W_{mp} to vary. The values obtained were $C_{DA} = (2.8 \pm 0.7) \times 10^{-40} \text{ cm}^6/\text{s}$ and $W_{\text{mp}}^{\text{exp}} = 6700 \text{ s}^{-1}$. The multiphonon decay rate was also estimated using the experimental decay rate obtained for the lowest Nd^{3+} doped sample together with the radiative decay rate (obtained by JO theory) by means of the equation $W_{\text{mp}}^{\text{cal}} = W_{\text{exp}} - W_{\text{rad}}$, resulting in $W_{\text{mp}}^{\text{cal}} = 8960 \text{ s}^{-1}$. The difference between $W_{\text{mp}}^{\text{cal}}$ and $W_{\text{mp}}^{\text{exp}}$ of 2260 s^{-1} , however, must be attributed to the energy transfer rate in this lowest Nd^{3+} doped sample. The distribution of η as a function of concentration (dots) and the fit (solid line) with Eq. (10) is presented in Fig. 7. A little similar to the Nd:PLZT case,¹⁵ the behavior of η versus Nd^{3+} concentration is strongly dependent on the C_{DA} value and slightly on the C_{DD} one, i.e., with a large variation of C_{DD} no appreciable change is observed in the fitting behavior, whereas a slight variations of C_{DA} varies appreciably the curve. This gives us clear indication that the highest influence on fluorescence quenching is mainly due to the CRs, although C_{DD} be almost one order of magnitude larger than C_{DA} . This way, by fixing the value of C_{DD} and letting C_{DA} and W_{mp} vary during the fitting procedure of the curve in Fig. 7, is a good method to obtain C_{DA} .

Since the reduction of η is balanced by the increase of the optical absorption coefficient with the increasing concentration, the figure of merit, defined as ηN_t versus N_t , is a way to determine optimum concentrations for practical applications. Figure 7 also presents the figure of merit that shows a maximum around $N_t = 6.0 \times 10^{20} \text{ Nd}^{3+} \text{ ions/cm}^3$, concentration in which $\eta = 0.24$. This concentration is almost twice that for Nd^{3+} -doped yttrium aluminum garnet (Nd:YAG) system,³⁵ and the $\eta = 0.24$ value is not so small when compared to other laser materials with similar Nd^{3+} concentration.^{9,28} In other words, the NYCaB100x system has a much reduced concentration quenching effect, which put it as a prospective material for future applications.

IV. CONCLUSIONS

In summary, in this work we presented a complete and comprehensive study of the spectroscopic and thermo-optical properties of Nd³⁺-doped yttrium calcium borate glasses (Nd:YCaB) with composition $x\text{Nd}_2\text{O}_3-(5-x)\text{Y}_2\text{O}_3-40\text{CaO}-55\text{B}_2\text{O}_3$ ($0 \leq x \leq 1.0$ mol%). The spectroscopic properties obtained by luminescence and absorption measurements together with the Judd–Ofelt analysis testified the higher asymmetry of the ligand field around the Nd³⁺ ions, which explains the relatively high radiative rate (low radiative lifetime) of the ⁴F_{3/2} Nd³⁺ level in this system. The thermal lens technique was used to quantify the fluorescence quantum efficiency (η) and the thermo-optical properties such as thermal diffusivity (D) and conductivity (K) and temperature coefficient of the optical path length change (ds/dT). The values obtained for D , K , and ds/dT are very similar to those of commercial laser glasses. A maximum $\eta = 0.54$ for the lowest doped sample was found to be substantially higher than for other borates glasses and also crystal with comparable Nd³⁺ content. The η dependence on the Nd³⁺ concentration was analyzed in terms of energy transfer processes based on the Förster–Dexter model of multipolar ion–ion interactions. The obtained energy transfer microparameters C_{DD} and C_{DA} and the weak dependence of η on the Nd³⁺ concentration with a high optimum Nd³⁺ concentration, as indicated by the figure of merit, put this system as a strong candidate for photonics applications.

ACKNOWLEDGMENTS

The authors are thankful to the Brazilian agencies CAPES, FAPESP (Grant No. 04/00093–0), CNPq, FINEP, and FAPEAL and to the Brazil–France CAPES–COFECUB (Grant No. 455/04) agreement for the financial support of this work. The research of D. R. S. Santos is supported by undergraduate studentships from PIBIC/CNPq and those of W. F. Silva and W. Q. Santos by graduate studentships from CAPES and CAPES/FAPEAL (Project No. PEB-2009–03-015 (04)), respectively.

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